

Department of Chemistry, Gargaon College

Sessional Exam, 1st Sem (SEC) SEC-01

Time: 1.5 hrs

Marks: 40

1. Who discovered the term Chromatography? 1
2. Answer the following 1
 - a) Mathematical expression for pH is _____
3. Why Chromatography is important? 2
4. How would you determine the detection of foreign resin in asafoetida 2
5. What is stationary phase and mobile phase in Chromatography? 2
6. What is the role of polar and non-polar solvent in determining the retention time in Chromatography? 3
7. What the procedure for measuring dissolved oxygen for a water sample. 3
8. Answer **any two** of the following: 2x2=4
 - a) Write about the sources responsible for contaminating water.
 - b) Write about the methods for purification of water.
 - c) Prepare 100 ml 0.05 N NaOH solution.
9. Answer **any three** of the following questions briefly: 3x2=6
 - a. Define food processing? What are the objectives of food processing?
 - b. What is adulteration? What are the detection techniques to determine of adulteration in coffee?
 - c. What are the adulterants present in a) Coriander powder b) Hing c) Chilli powder
 - d. Why should you not buy shining pulses from the market?
10. Answer any four of the following. 2x4=8
 - a) Define various methods used in analytical chemistry.
 - b) Write down the differences between qualitative and quantitative analysis.
 - c) What is concept of sampling? What are important terms involved with this?
 - d) Write down the differences between random and non- random sampling.
 - e) Define accuracy and precision with examples.
 - f) Write down four rules to find out significant figure.
11. Answer any two 4x2=8
 - a) What is soil pH? Why it is important? How can we change soil pH?
 - b) What is complexometric titration? What are the common indicators used for complexometric titration? Write basic principle of complexometric titration with EDTA.
 - c) How Calcium and magnesium ions present in soil sample can be determined by complexometric titration? Explain the principle and procedure.

*****ALL THE BEST*****